

# Determining Dilution Series

This technote provides information about the concentration range spanned and preparation calculations for serial dilutions. *Figure 1* illustrates the range of concentrations spanned by a series of twelve dilutions with commonly utilized dilution factors. The following also provides calculations for the preparation of a serial dilution factor. **Bold** indicates standard calculations regardless of the dilution factor used. Refer to *Figure 2* to clarify the experimental set up.

## Sample Experiment Example:

- 3 mL Sample Volume
- 2.5 fold serial dilution

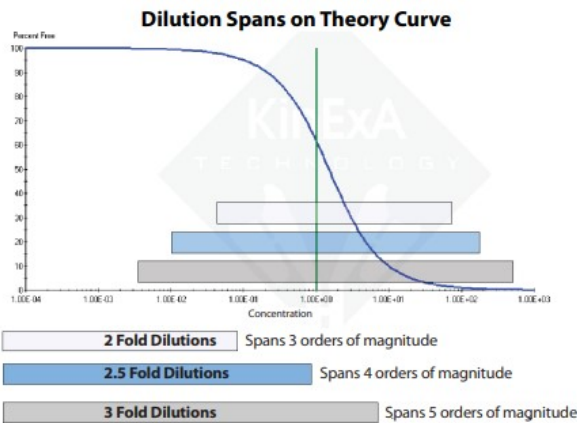


Figure 1.

## Prepare Serial Dilutions:

- Calculate the total *Sample Volume* necessary to run each sample.
- Calculate the *Starting Volume* for tube 1:

$$\text{Starting Volume} = \frac{\text{Sample volume} \times \text{Dilution Factor}}{(\text{Dilution Factor} - 1)}$$

$$\begin{aligned} \text{Starting Volume} &= (3 \text{ mL} \times 2.5) / (2.5 - 1) \\ \text{Starting Volume} &= 5 \text{ mL} \end{aligned}$$

- Calculate the *Pipet Transfer Volume*:

$$\text{Pipet Transfer Volume} = \text{Starting Volume} - \text{Sample Volume}$$

$$\begin{aligned} \text{Pipet Transfer Volume} &= 5 \text{ mL} - 3 \text{ mL} \\ \text{Pipet Transfer Volume} &= 2 \text{ mL} \end{aligned}$$

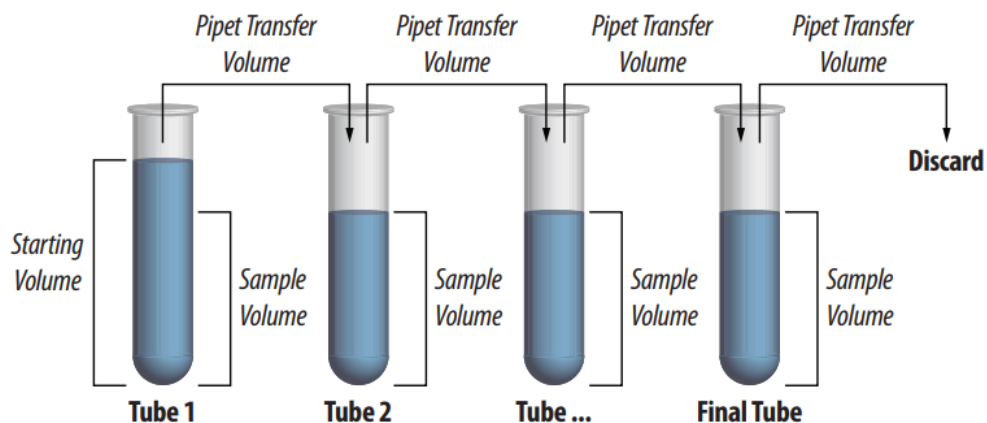


Figure 2.